## Biomineral-inspired Materials Chemistry Episode 2

Inorganic-Organic composites

> Nanomaterials

Funsctional materials and interfaces

> Oriented crystals

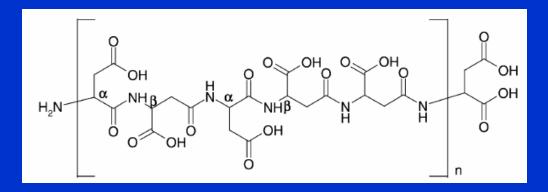
> Materials with complex morphologies

- Organized assemblies
- > Hierarchical materials

#### **Biomineral matrices**

Extraction of organic matrices and re-use for re-mineralization

Reactivation of chitin hydrophobic domains with polyaspartate to lead to oriented growth of calcite



polyaspartate

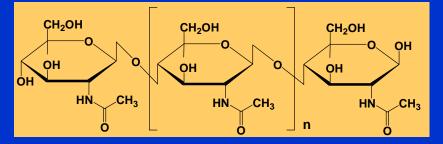
#### Pure $\beta$ -chitin from cuttlebone as a matrix

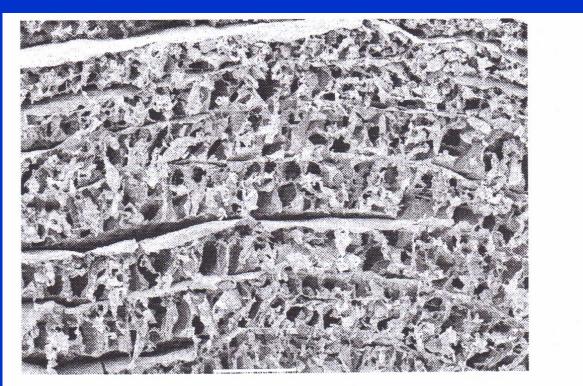




Intact sponge-like  $\beta$ -chitin matrix from demineralized cuttlebone. Scale bar, 500  $\mu$ m.

# Silica replicas can be obtained from β-chitin from cuttlebone





Silica replica of the  $\beta$ -chitin matrix of cuttlebone. Scale bar, 500  $\mu$ m.

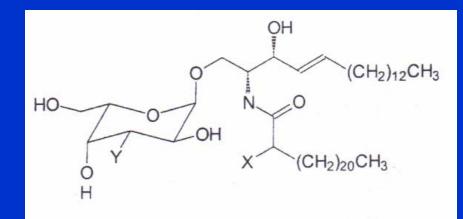
#### **Lipid Tubules**

Biomineral organic matrices are not easy to obtain in large amounts, so synthetic analogs with functionalized surfaces are often used for inorganic nucleation

Lipid tubules are multi-lamellar structures formed by the supramolecular assembly of chiral amphi-philic molecules

In the initial stages molecules pack in bilayer sheets separated by solvent. But the molecular chirality induces formation of long strings of strongly-interacting chiral amphiphiles.

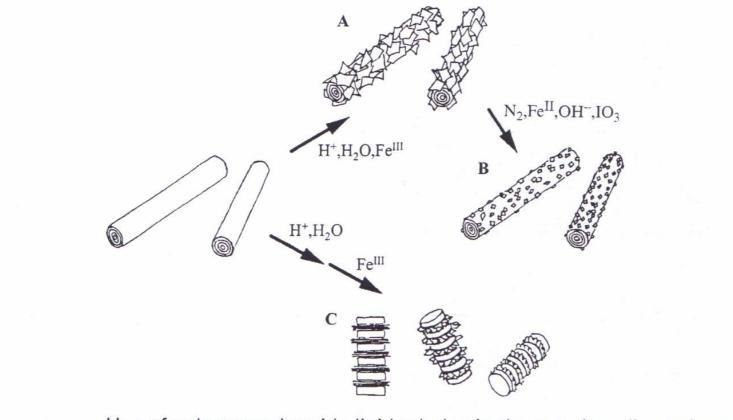
## **Lipid Tubules**



X = OH	Y = OH	HO - Cer
X = H	Y = OH	H – Cer
X = OH, H	$Y = OSO_3^-$	S – Cer

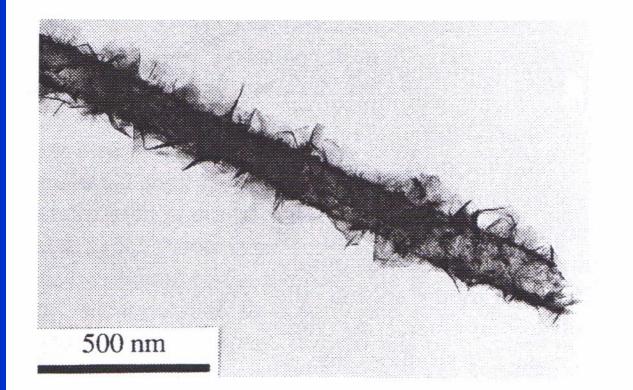
Molecular structure of galactocerebroside lipid and three derivatives.

## Formation of rod-like iron oxide composites



Use of galactocerebroside lipid tubules in the template-directed synthesis of iron oxides.

## **FeOOH formation on lipid tubule**

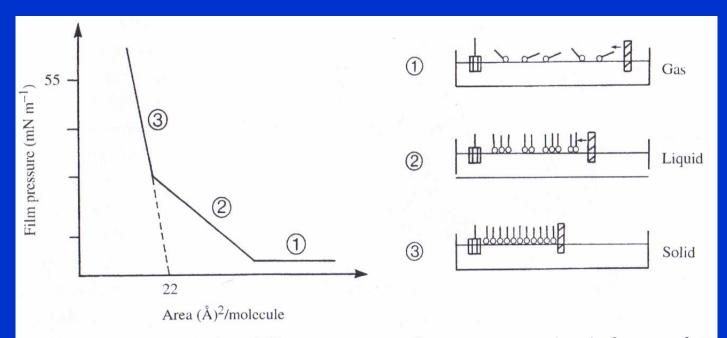


Galactocerebroside lipid tubule coated with lepidocrocite (y-FeOOH).

#### **Oriented formation on soap films**

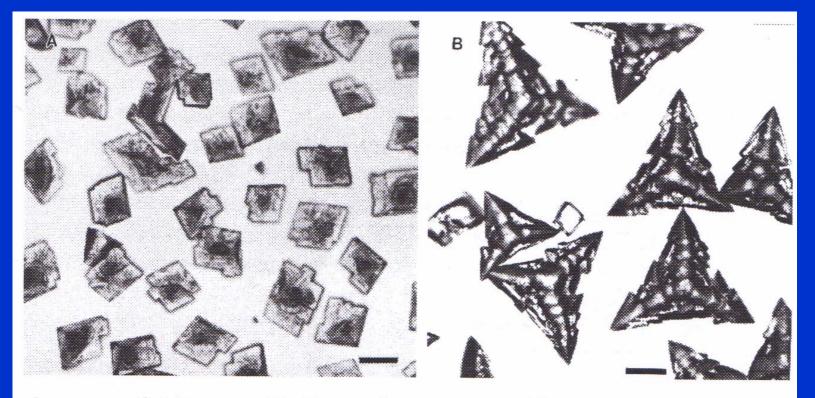
Controlled crystallization of inorganic solids on compressed monomolecular films of insoluble surfactants, spread at the air-water interface

#### Langmuir monolayers



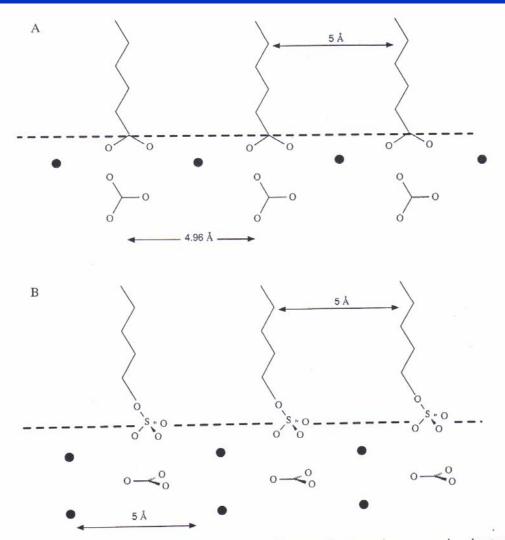
Idealized plot of film pressure against area per molecule for a surfactant undergoing compression at the air-water interface. The corresponding gas, liquid and solid states of the monolayer are also shown.

## Formation of calcium carbonate under Stearic acid Langmuir monolayers



Calcite crystallization under compressed Langmuir monolayers: (A) stearic acid film with  $\{1\overline{1}0\}$  oriented crystals, scale bar, 50  $\mu$ m; (B) *n*-eicosyl sulfate film with  $\{001\}$  oriented crystals, scale bar, 20  $\mu$ m.

### **Mechanism of formation**

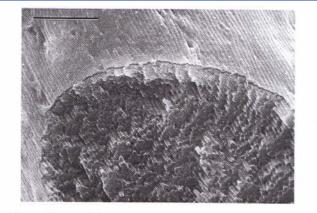


**Fig. 9.22** Interfacial recognition under Langmuir monolayers and oriented calcite nucleation for: (A) carboxylate monolayers and the {110} face; (B) sulfate monolayers and the {001} face.

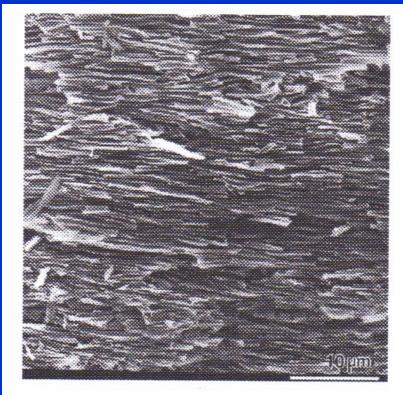
#### **Morphosynthesis of biomimetic form**

One major challenge in biomineral-inspired materials chemistry is the synthetic reproduction of analogous structures, using an approach called MORPHOSYNTHESIS

# Physical patterning with supramolecular templates

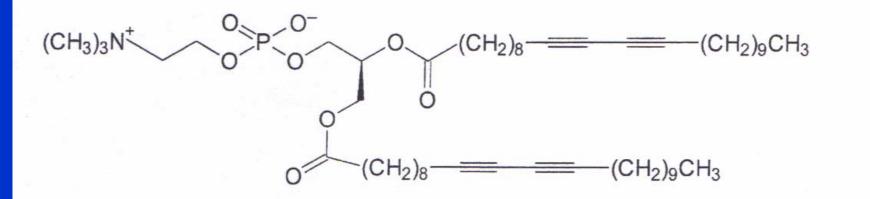


Sectioned bacterial thread showing internal hexagonal superstructure of coaligned multicellular filaments. Scale bar, 10  $\mu$ m.



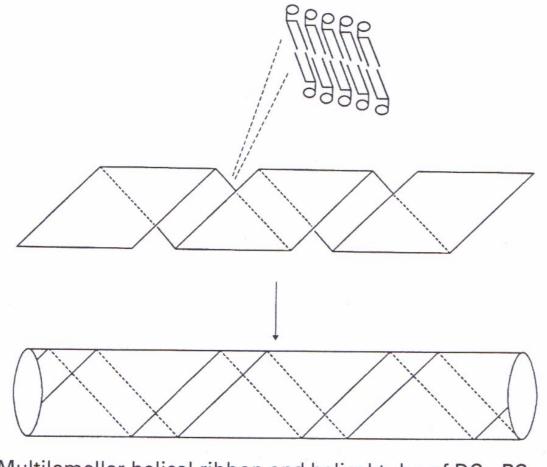
Section cut parallel to the long axis of a silica-infiltrated bacterial thread after removal of the multicellular filaments, showing coaligned channels. Scale bar, 10  $\mu$ m.

### Synergism in the assembly of DCPC



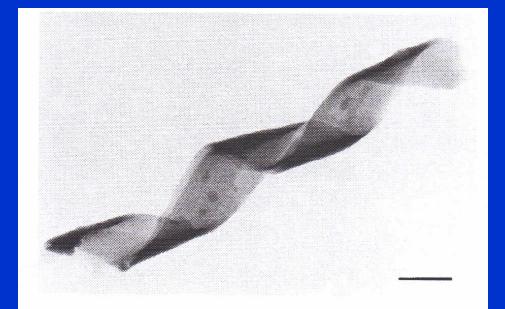
Molecular structure of diacetylenic phosphatidyl choline (DC<sub>8,9</sub>PC)

### **Multilamellar helical ribbons**



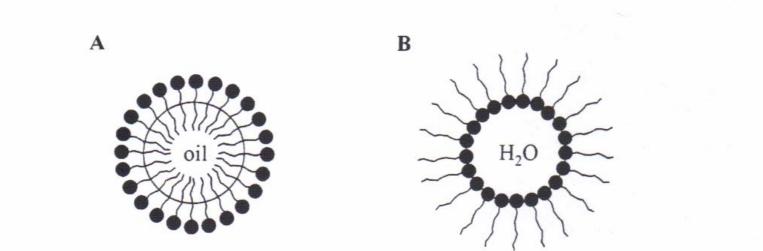
Multilamellar helical ribbon and helical tube of  $DC_{8,9}PC$ .

## Silica helical ribbon



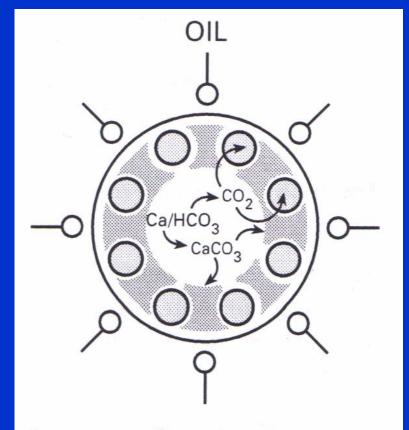
Silica-lipid mineralized helical ribbon. Scale bar, 400 nm.

#### **Inorganic solids formed in reverse microemulsions**



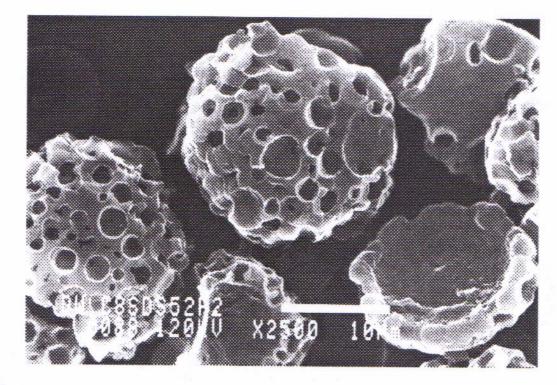
(A) Oil-in-water microemulsions; (B) water-in-oil reverse microemulsions

## **CaCO<sub>3</sub>** formed in reverse microemulsions



Growth and patterning of calcium carbonate hollow shells in reverse microemulsions.

### **CaCO<sub>3</sub>** formed in reverse microemulsions



Calcium carbonate hollow shells with surface pores. Note also the presence of a broken shell. Scale bar, 10  $\mu$ m.